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INTERNATIONAL PRELIMINARY EXAMINATION REPORT

(PCT Article 36 and Rule 70)

Applicant's or agent's file reference M80580405:BGC:DVL	FOR FURTHER ACTION	See Notification of Transmittal of International Preliminary Examination Report (Form PCT/IPEA/416).				
International Application No.	International Filing Date (day/month/year)		Priority Date (day/month/year)			
PCT/AU2003/001336	10 October 2003		10 October 2003			
International Patent Classification (IPC) or	International Patent Classification (IPC) or national classification and IPC					
Int. Cl.			·			
C08K 3/20 (2006.01) C08K 3/22 (200 C08K 3/14 (2006.01) C08K 3/30 (200						
Applicant.						
COMMONWEALTH SCIENTIF	IC AND INDUSTRL	AL RESEARCH O	RGANISATION et al			
1. This international preliminary examination report has been prepared by this International Preliminary Examining Authority and is transmitted to the applicant according to Article 36.						
2. This REPORT consists of a total of 3	sheets, including this c	over sheet				
			claims and/or drawings which have been			
amended and are the basis for this 70.16 and Section 607 of the Adn	s report and/or sheets co	ntaining rectification	s made before this Authority (see Rule			
·•		under the PC1).				
These annexes consist of a total o	f 5 sheet(s).					
3. This report contains indications relating	to the following items:					
I X Basis of the report	·					
II Priority		.`				
III Non-establishment of opi						
IV Lack of unity of invention						
Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement						
VI Certain documents cited	•					
VII Certain defects in the inte	mational application	•	•			
VIII Certain observations on ti	ne international applicat	ion				
Date of submission of the demand	Γ.					
10 May 2005	Date of completion of the report 31 January 2006					
Name and mailing address of the IPBA/AU		Authorized Officer				
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INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No.
PCT/ATI2003/001336

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INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No. PCT/AU2003/001336

V. Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

1.	Statement

Novelty (N)	Claims	1-49	YES
	Claims		NO
Inventive step (IS)	Claims	12-49	YES
	Claims	1-11	NO
Industrial applicability (IA)	Claims	1-49	YES
	Claims	•	NO

2. Citations and explanations (Rule 70.7)

D1 WO 2001057142

D2 EP 1125969

D3 EP 292120

D4 EP 345644

D5 EP 486225

D6 EP 506033

New Citations

D7 US 6203906 (Christie et al) 20 March 2001

D8 US 6046267 (Vidaurre et al) 4 April 2000

NOVELTY (N) Claims 1-49

D1-D7 disclose polymeric films and resins with inorganic filler particles in the size range and loadings claimed. The citations do not explicitly disclose a polymer article that is substantially free of gas bubbles or the methods of preventing gas bubbles claimed. Claims 1-49 are therefore novel.

INVENTIVE STEP (IS) Claims 1-11

Claims 1-11 do not require the polymeric articles to be made free of gas by any particular method.

D1-D7 disclose polymeric films and resins with inorganic filler particles in the size range and loadings claimed.

D8 discloses a method of degassing polymeric composites that are reinforced with particulate filler under reduced pressure conditions. To overcome problems with gas bubbles in reinforced polymeric product it would be obvious to use a reduced pressure degassing step in the preparation process as taught by D8. The use of particular filler particle types or size sub-ranges or restricting the polymer used to thermoplastic polymers are not considered pertinent to the question of inventive step. Claim 10 refers to the use of a hydrophobicity matching dispersion agent but this feature only appears to have significance in reducing bubble content when the particles are required to be milled under low shear conditions. For these reasons claims 1-11 lack an inventive step.

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CLAIMS

- 1. A polymer article comprising a thermosetting polymer matrix and inorganic ultrafine particulate material evenly distributed through the polymer matrix characterised in that the inorganic particulate material has a particle size up to 10 μ m at a particle loading of 0.01 20 wt% based on the total weight of the polymer inorganic components, wherein the polymer article is substantially free of gas bubbles.
- 2. The article of claim 1, wherein the particle size is in the range of 1 to 1000 nanometers.
- 10 3. The article according to claim 1, wherein the upper size limit of the particle is less than 800 nanometers.
 - 4. The article according to claim 1, wherein the particulate material has a size range of between 100-800 nanometers.
- The article according to any one of claims 1-4, wherein the upper particle
 loading limit is 10 wt% based on the total weight of polymeric and inorganic material.
 - 6. The article of claim 1, wherein the interparticle distance between the particles is less than 20 μ m and the average interparticle distance is in the range of 2 μ m to 10 μ m.
- 7. The article according to claim 1, wherein the inorganic particulate material is an inorganic material available in submicron powder form, or a precursor to said inorganic particulate material, selected from the group including metal oxides, metal silicates and metal alkoxides.
- 8. The article according to claim 7, wherein the particulate material is an oxide, silicate and alkoxide of aluminium, titanium and silicon.
 - 9. The article according to claim 7, wherein the particulate material is selected from the group consisting of alumina, (Al_2O_3) , titania (TiO_2) , zirconia (ZrO_2) , silica (SiO_2) , silicon carbide, hydroxides of alumina and alumino- silicates.
- 10. The article according to claim 1, wherein the particulate material is coated or treated with a dispersing agent to match the hydrophobicity of the particles with

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the polymer.

- 11. The article according to claim 1, wherein the polymeric material is a polyurethane, polyester, hybrid or copolymer thereof.
- 12. A method of forming a polymer-inorganic composite article comprising the5 steps of:

dispersing an inorganic submicron particulate material having a particle size up to 10 µm in a polymer at a loading rate of 0.01 to 20 wt% of the total weight of polymer and inorganic material, the dispersion step preventing the formation of gas bubbles in the mixture to provide a dispersion having a substantially even distribution of particulate material in the polymer; and

casting the dispersion in a mould to cure the mixture.

- 13. The method of claim 12, wherein the particle size is in the range of 1 to 1000 nanometers.
- 14. The method according to claim 13, wherein the mixing is carried out to minimise the amount of agglomeration of dispersed particles and provide a substantially even distribution of particles in the polymer.
 - 15. The method according to claim 14, wherein the mixing step includes milling the particulate material with the polymer under sub-atmospheric conditions.
- 16. The method of claim 15, wherein the mixing is carried out in a low shear regime having a Reynolds number less than or equal to 1000.
 - 17. The method of claim 15, wherein the mixing is carried out in a high shear regime having a Reynolds number much greater than 1000.
 - 18. The method of claim 16, wherein the particulate material is coated or treated with a dispersing agent to match the hydrophobicity of the particles with the polymer.
 - 19. The method of claim 18, wherein the dispersing or compatibilizing agent is selected from the group consisting of non-alcohol or non amine aromatic solvent.
 - 20. The method of any one of claims 12 to 19, wherein the mixing occurs at under vacuum conditions to ensure no bubbles are formed in the dispersion.

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- 21. The method of any one of claims 12 to 19, wherein the mixing occurs under a pressure of less than one atmosphere to ensure no bubbles are formed in the dispersion
- The method of claim 20 or 21, wherein the mixing is conducted in a vacuum
 oven at a temperature above melting temperature and below the decomposition temperature of the polymer.
 - 23. The method of any one of claims 12 to 22, wherein the mixing takes place in a vacuum up to 760 mm Hg with the level dependent on the viscosity of the dispersion.
- 10 24. The method of claim 16, wherein the particulate material is mixed with a solvent prior to mixing with the polymer material, with the particulate material, the polymer material and the solvent forming a slurry.
- The method of claim 24, wherein the solvent is removed by evaporation from the dispersion and the dispersion is cured to form a composite substantially
 free of gas bubbles.
 - 26. The method of claim 17, wherein the inorganic particulate material is dispersed in the polymer by milling the particulate material and polymer at a mill temperature, above the melting temperature resin but below the decomposition temperature of the polymer resin at a pressure below atmospheric pressure.
- 20 27. The method of any one of claims 12 to 26, wherein after the mixing step polymerisation of the polymer is initiated by the addition of an initiator and the dispersion cured.
- 28. The method of any one of claims 12 to 26, wherein at least one other additive selected from the group consisting of anti static agents, fillers, pigments,
 25 optical brighteners and UV brighteners are added to the dispersion.
 - 29. A method of dispersing an inorganic particulate material substantially homogenously in a polymer, the particulate material consisting essentially of particles having a particle size up to 10 µm at a loading rate in the range of 0.01 to 20 wt% based on the total weight of the particulate and polymer material, the method comprising the steps of combining the particulate material and a powder of the polymer under sub atmospheric pressure conditions, the combination being

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milled at a temperature above the melting temperature but below the decomposition temperature of the polymer to form a dispersion, and curing the dispersion.

- 30. The method of claim 29, wherein the particle size is in the range of 1 to 1000 nanometers.
- 31. The method of claim 30, wherein the particulate material is added to a solvent carrier prior to addition to the polymer to form a slurry, the solvent carrier being evaporated from the dispersion.
- 32. The method of claim 30 or 31, wherein the milling is conducted at a pressure of typically less than 900 millibar (gauge).
 - 33. A method of increasing the wear resistance of a polymer comprising the steps of

dispersing an inorganic submicron particulate material having a particle size up to 10 µm in a polymer at a loading rate of 0.01 to 20 wt% of the total weight of polymer and inorganic material, the dispersion step preventing the formation of gas bubbles in the mixture to provide a dispersion having a substantially even distribution of particulate material in the polymer; and

casting the dispersion in a mould to cure the polymer

- 34. The method of claim 33, wherein the particle size is in the range of 1 to 1000 nanometers.
 - 35. The method according to claim 34, wherein the dispersing step is carried out to minimise the amount of agglomeration of dispersed particles and provide a substantially even distribution of particles in the polymer.
- 36. The method according to claim 35, wherein the mixing step includes milling the particulate material with the polymer under vacuum conditions.
 - 37. The method of claim 36, wherein the mixing is carried out in a low shear regime having a Reynolds number less than or equal to 1000.
 - 38. The method of claim 36, wherein the mixing is carried out in a high shear regime having a Reynolds number much greater than 1000.

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- 39. The method of claim 37, wherein the particulate material is coated or treated with a dispersing agent to match the hydrophobicity of the particles with the polymer.
- 40. The method of claim 39, wherein the dispersing agent is PGMA.
- 5 41. The method of any one of claims 33 to 40, wherein the mixing occurs at under vacuum conditions to ensure no bubbles are formed in the dispersion.
 - 42. The method of any one of claims 33 to 40, wherein the mixing occurs under a pressure of less than one atmosphere to ensure no bubbles are formed in the dispersion
- 10 43. The method of claim 41 or 42, wherein the mixing is conducted in a vacuum oven at a temperature above melting temperature and below the decomposition temperature of the polymer.
 - 44. The method of any one of claims 33 to 43, wherein the mixing takes place in a vacuum up to 760 mm Hg with the level dependent on the viscosity of the dispersion.
 - 45. The method of claim 37, wherein the particulate material is mixed with a solvent prior to mixing with the polymer material, with the particulate material, the polymer material and the solvent forming a slurry.
- 46. The method of claim 45 wherein the solvent is evaporated from the dispersion and the dispersion is cured to form a composite substantially free of gas bubbles.
 - 47. The method of claim 38, wherein the inorganic particulate material is dispersed in the polymer by milling the particulate material and polymer at a mill temperature, above the melting temperature but below the decomposition temperature of the polymer resin at a pressure below atmospheric pressure.
 - 48. The method of any one of claims 33 to 47, wherein after the mixing step polymerisation of the polymer is initiated by the addition of an initiator and the dispersion cured.
- 49. The method of any one of claims 33 to 47, wherein at least one other additive selected from the group comprising anti static agents, fillers, pigments, optical brighteners and UV brighteners are added to the dispersion.

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